

Roll No.

Total No. of Pages : 02

Total No. of Questions : 09

B.Tech. (Chemical Engg.) (Sem.-4)

MASS TRANSFER-I

Subject Code : BTCH-402B

M.Code : 78132

Date of Examination : 05-07-22

Time : 3 Hrs.

Max. Marks : 60

INSTRUCTIONS TO CANDIDATES :

1. SECTION-A is COMPULSORY consisting of TEN questions carrying TWO marks each.
2. SECTION-B contains FIVE questions carrying FIVE marks each and students have to attempt any FOUR questions.
3. SECTION-C contains THREE questions carrying TEN marks each and students have to attempt any TWO questions.

SECTION-A

1. Write briefly :

- a) Define molecular diffusion.
- b) Give the physical significance of the dimensionless groups in mass transfer.
- c) Define Fick's first law of diffusion.
- d) What is the physical significance of HTU?
- e) What is Reflux Ratio?
- f) What is batch distillation?
- g) List out the assumptions for McCabe Thiele method?
- h) What are the common packings and materials for a cooling tower?
- i) What is equimolar counter diffusion?
- j) What is Raoult's law?

SECTION-B

- In an oxygen - nitrogen mixture at 10 atm and 25°C, the concentration of oxygen at two places of 2mm apart are 10 and 20 volume percent respectively. Calculate the rate of diffusion of oxygen expressed as gm/cm².hr for the case of unicomponent diffusion (nitrogen is non-diffusing). Value of diffusivity is 0.181 cm²/sec.
- Briefly explain the penetration theory of mass transfer coefficient.
- Explain loading and flooding in packed towers.
- Explain properly with diagram, the drying rate curve.
- It is desired to separate a feed mixture containing 40% heptane and 60% ethyl benzene, such that 60% of the feed is in distillate. Estimate the composition of the bottoms and distillate when the process distillation is equilibrium distillation. Given equilibrium data :

$x :$	0	0.08	0.185	0.251	0.335	0.489	0.651	0.79	0.914	1.0
$y :$	0	0.233	0.428	0.514	0.608	0.729	0.814	0.91	0.963	1.0

x : mole fraction of heptane in liquid phase and

y : mole fraction of heptane in vapour phase.

SECTION-C

- Differentiate packed towers with tray towers.
- Discuss the criteria for choice of solvent for liquid - liquid extraction.
 - Explain with neat diagram the material balance for the multi stage liquid - liquid extraction.
- Explain the characteristics of azeotropes with T-x-y and P-x-y diagrams.

NOTE : Disclosure of Identity by writing Mobile No. or Making of passing request on any page of Answer Sheet will lead to UMC against the Student.